

Dalton's table of the tension, or elastic forces of vapour at different temperatures, it appears that the tension of vapour at  $32^{\circ}$  is equal to the weight of  $\cdot 200$  inch of mercury; and that the difference between the tension of vapour at  $32^{\circ}$  and the tension of vapour at  $33^{\circ}$ , the value, namely, of the first term or unit, in our assumed arithmetical series, is  $\cdot 007$  inch of mercury. Now, the difference between  $32^{\circ}$ , and  $80^{\circ}$ , the mean temperature at the level of the sea under the Equator, is  $48^{\circ}$ . Supposing, therefore, each of these 48 degrees to increase in an arithmetical progression,  $\cdot 007$  for each degree; the tension for the whole 48 degrees will amount to  $\cdot 336$ ; which tension added to  $\cdot 200$ , the tension at  $32^{\circ}$ , gives  $\cdot 536$  inch, as the tension corresponding to the vapour at  $80^{\circ}$ , the temperature of the earth's surface under the Equator. But, by Dr. Dalton's same table of tensions, we find that  $\cdot 536$  does not represent the proper tension of vapour at  $80^{\circ}$ , but of vapour at about  $61^{\circ}$  only. According to this estimate it follows, that at the Equator, while the temperature of the air over the earth's surface is  $80^{\circ}$ , the point of saturation with vapour is  $19^{\circ}$  below that temperature. Hence, at the Equator, the air immediately incumbent on the earth's surface must be comparatively dry. Moreover, the cause which has been thus shown to produce the dryness of the Equatorial air, at the earth's surface, must all